

# **MANGANESE**

#### **MANGANESE IN SOILS**

Manganese (Mn) is the second most abundant trace element in the earth's crust after iron. The concentration of manganese is soils typically exceeds that of the macronutrients such as phosphorus and sulphur and often that of nitrogen. Consequently, where plant deficiencies occur on soils with adequate or high total manganese, it is because most of the Mn is present in forms that are not available for plant uptake. The total amount of manganese in soils is typically around 0.25% and is normally in the range of 0.02 – 1%. It can be as high as 13% in some volcanic soils.

Plants take up manganese from the soil as the divalent manganese ion Mn<sup>2+</sup>, however several factors can affect manganese availability.

## Soil pH

Manganese availability decreases as the pH increases. Above a pH<sub>water</sub> of 7.5, manganese availability may not be adequate to meet plant demand. At pH<sub>water</sub> levels below 5.5, manganese becomes very soluble, and toxicity may occur. Toxicity is usually associated with other acidic soil infertility problems such as aluminium toxicity and deficiencies of calcium, magnesium, and molybdenum.

# Soil organic matter

Complexes form between manganese ions and organic matter in alkaline soils that are high in organic matter, reducing the amount of Mn<sup>2+</sup> available for plant uptake.

#### Soil disturbance

Cultivation will increase the availability of manganese in the soil by accelerating the decomposition of soil organic matter.

#### The weather

Exchangeable manganese levels in the soil can fluctuate depending on seasonal conditions. Cold wet conditions sometimes induce deficiency, possibly attributable to the combined effects of reduced mineralization of soil organic matter and reduced root growth and reduced metabolic activity in roots affecting manganese uptake.

#### Soil moisture

Under water-logged conditions, insoluble manganese oxide can be reduced by soil bacteria to Mn<sup>2+</sup>. This may result in temporary toxicity. Under very dry conditions, insoluble dehydrated manganese salts can form in the soil, reducing the availability of manganese.

#### MANGANESE IN PLANTS

Once taken up and incorporated into plant tissue, manganese is relatively immobile in the plant and is not readily relocated from old to young tissue. High concentrations of other cations in the soil solution

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e.g., calcium, zinc, magnesium, and ammonium may reduce manganese uptake by plants. Conversely, manganese may depress the uptake of other cations such as iron e.g., in pineapples and ginger, resulting in iron deficiency.

Manganese deficiency occurs in plants grown in alkaline soils but is not common elsewhere. Toxicity occurs on very acid, poorly drained and water-logged soils.

## **DEFICIENCY SYMPTOMS**

Manganese deficiency symptoms appear first in young leaves, with symptoms closely resembling those of magnesium deficiency. In both cases, interveinal chlorosis (yellowing) occurs in the leaves. However, magnesium deficiency appears first on older leaves. In cereals and grasses, greyish or brownish spots and streaks occur in the middle or basal parts of younger leaves. These necrotic spots may merge into a band across the leaf, isolating the still green end portion of the leaf.

Manganese deficiency symptoms in broad-leaf plants (dicotyledons) occur as small yellow spots on the younger leaves, which turn brown or black. The abscission of developing leaves commonly occurs, leading to reduced flower formation.

In tree crops, deficiency symptoms usually appear on recently matured leaves during early summer growth. This is different to the development on very young leaves in the case of iron deficiency or, old leaves in the case of magnesium and potassium deficiency. Leaf shape and size, and shoot length are usually normal but symptoms are generally greater on the southern or shady side of trees (in the southern hemisphere).

## **TOXICITY SYMPTOMS**

Manganese toxicity is characterised by:

- raised interveinal areas giving a puckered appearance,
- red, brown, or black spotting of the older leaves,
- an uneven distribution of chlorophyll.

If the toxicity continues, the plants will wilt and die prematurely. Plants particularly susceptible to manganese toxicity are lucerne, cabbage, cauliflower, cereals, clover, pineapple, potato, and tomato.

# **MANGANESE FERTILISERS**

Incitec Pivot markets a granular grade of Manganese Sulphate for dry application to the soil. It is primarily intended for use in planting fertilisers for grain crops in in soils derived from limestone in South Australia. Its effectiveness elsewhere is likely to be limited e.g., soil application in tree crops.

Soluble grades of manganese sulphate and manganese chelate may be applied in solution e.g., foliar sprays. Chelate formulations are subject to less soil fixation but are more costly.

#### **APPLICATION**

Soil applied manganese can be rapidly fixed or converted to forms that are unavailable for plant uptake. For this reason, foliar sprays are generally recommended. One of the few situations where soil applications are used is at planting in grain crops on calcareous soils in South Australia. Banding

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the manganese with the seed is more effective than pre-plant broadcast applications as it exposes the manganese to less soil, reduces fixation and allows lower rates to be used.

Foliar sprays are suited to horticultural crops. In tree crops, annual foliar applications in spring or as soon as there is a good cover of new leaves, is usually sufficient. If the deficiency is severe more than one spray may be required. Manganese will not need to be applied in fertiliser programs where manganese-based fungicides such as Mancozeb e.g., Dithane M45, are applied on a routine basis in horticultural crops.

#### **FURTHER READING**

Seek professional advice before use. A more detailed Agritopic is available from Incitec Pivot on Manganese.

#### SAFETY DIRECTIONS

Refer to the Safety Data Sheet (SDS) for more detailed safety advice. Before use, read the Product Label and the SDS. Use safe work practices and avoid contact with the eyes and skin. Avoid ingestion and inhaling dust. Protective clothing, eyewear and dust masks should always be used when dealing with this product. Observe good personal hygiene, including washing hands after use. Avoid loss of fertiliser to waterways.

## **WARNING**

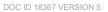
This document contains information of a general nature. Before using fertiliser seek independent agronomic advice. Fertiliser programs may need to be varied depending on the plants being grown, climatic and soil conditions, application methods, irrigation, agricultural and livestock management practices, the soil's fertility, and cultural practices. ('Unforeseen Elements')

Fertiliser may burn and/or damage crop roots or foliage. Foliar burn to the leaves, fruit or other plant parts is most likely to occur when fertilisers are foliar applied at high concentrations and/or on a regular basis, different products are mixed and sprayed together at cumulatively high rates, the water is of poor quality, or the spray is applied under hot dry conditions, e.g. in the heat of the day.

Fertiliser and supplements may affect animal health. Seek independent advice before using any supplements in livestock rations.

#### **DISCLAIMER**

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